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POGIL™ Activities for High School Chemistry6. ... Explain your answer. Oxygen. It is the second element in the compound. The elements are covalently bonded so a prefix is needed to show how many atoms of the element are in the molecule. Name the molecule NO. Nitrogen monoxide. ... 18 Naming Molecular Compounds-S

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Assuming that the name of the compound gives a clue to its molecular formula, predict how many atoms each of these prefixes indicates, and provide two examples. mono- one carbon monoxide or chlorine monofluoride di- two carbon dioxide or dinitrogen pentoxide penta- five chlorine pentafluoride or phosphorus pentachloride Naming Molecular Compounds 1 Model 2 - Prefixes and Suffixes Prefix ...

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naming molecular compounds pogil answer key Media Publishing eBook, ePub, Kindle PDF View ID 543babdb8 May 06, 2020 By Jackie Collins name or symbol of the metal comes first 6 use the table of ions in model 1 to answer the following

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Al is a metal, so the elements are held together by ionic bond. No prefixes are used when naming ionic compounds boron is a metalloid and non-metal, so forms covalent bonds with non-metals to make a molecules Naming Molecular Compounds 5 21. In the table below, first identify the type of bonding present in each compounds.

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4 POGIL™ Activities for High School Chemistry 16. For each of the following compounds, indicate whether or not your naming rules from Question

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15 will apply. If not, explain why the naming rules do not apply. FeI. 3. ICl. 5. HBrO. 4. 17. Using the rules your group developed in Question 15, name each of the following molecular compounds ...

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Molecular Geometry Pogil Answer Key

18. A student does not “waste” his time drawing a Lewis structure before determining the shape of PF₃. The student thinks that the shape of PF₃ must be trigonal planar because there are three fluorine atoms bonded to the central phosphorus atom. a. Draw the Lewis structure for PF₃. b. Was the student’s answer for the shape of a PF₃ ...

20 Molecular Geometry-S

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