

## **Chapter 19 Acids And Bases Study Guide For Content Mastery**

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## **Chapter 19 Acids And Bases**

An acid is a substance that contains hydrogen and ionizes to produce hydrogen ions in aqueous solution; A base is a substance that contains a hydroxide group and dissociates to produce a hydroxide ion in aqueous solution. Bronstred-Lowry model. A model of acids and bases in which an

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acid is a hydrogen-ion donor and a base is a hydrogen-ion acceptor.

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Chapter 19 - Acids & Bases Chapter 19 covers properties of acids and bases, the different definitions,  $[H^+]$  and  $[OH^-]$  and pH and pOH,  $K_a$  and  $K_b$  and their use when dealing with weak acids and bases and buffers.

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He said that acids are hydrogen containing compounds that ionize to yield hydrogen ions ( $H^+$ ) in aqueous solution. Also said that bases are compounds that ionize to yield hydroxide ions ( $OH^-$ ) in aqueous solution

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Answer. The Arrhenius definition states that acids are compounds that contain hydrogen, and when placed in aqueous solution, ionize to form hydrogen ( $H^+$  ions). Bases are compounds that ionize to yield hydroxide ions ( $OH^-$ ) in aqueous

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## **Chapter 19 Acids, Bases, and Salts**

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### SECTION 19.1 ACID-BASE

THEORIES (pages 587–593) This section compares and contrasts acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. It also identifies conjugate acid-base pairs in acid-base reactions.

## **Chapter 19: Acids and Bases**

### **Homework Packet (50 pts)**

Section 19.1 – Acid-Base Theories Acids have a sour taste, change the color of an indicator, can be strong or weak electrolytes in aqueous solution, and react with metals.

## **Chapter 19 - Acids, Bases, and Salts**

of acid or base are added. A buffer system should contain considerably more acid than base. Specific buffer systems should be chosen based on the pH that must be maintained. 114 Study Guide for Content Mastery Chemistry: Matter and Change Chapter 19

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Acids have a sour, tart taste, react with

metals such as Zn and Mg, react with

compounds containing Hydroxide ions

(good proton acceptors to form water

and a salt). Bases have a bitter taste,

feel slippery, reacts with an acid to form

water and a salt, and can form strong or

weak electrolytes in aqueous solution.

Work Step by Step



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## **Chemistry (12th Edition) Chapter 19 - Acids, Bases, and ...**

Arrhenius said that acids are hydrogen-containing compounds that ionize to yield hydrogen ions ( $H^+$ ) in aqueous solution. He also said that bases are compounds that ionize to yield hydroxide ions ( $OH^-$ ) in aqueous solution.

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634 Chapter 18 • Acids and Bases  
Section 118.18.1 Figure 18.1

Rhododendrons flourish in rich, moist soil that is moderately acidic, whereas *semper-vivum*, commonly called hen and chicks, grows best in drier, slightly basic soil. Introduction to Acids and Bases-!).)DEADifferent models help describe the behavior of acids and bases.

**Chapter 18: Acids and Bases**  
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